

**Using Materials Science
to Stimulate Students'
"Need to Know"**

Terre Trupp
RET I
Union-Endicott High School
Applied Chemistry

8/10/2002 T. Trupp

Project Goals

- Develop two laboratory activities designed to engage students, increase higher-order thinking skills, and acquaint students with experimental design
- Utilize everyday materials that relate more to students' learning needs

8/10/2002 T. Trupp

**The Four Aspects of
Materials Science**

- Synthesis and Processing
 - Preparative conditions that determine atomic structure/microstructure
- Structure and Composition
 - Arrangement and identity of atoms derived synthetically and used to prepare the material
- Properties
 - Characteristics of the material
- Performance
 - Conditions where material maintains desirable characteristics

8/10/2002 T. Trupp

**Using Toys to Teach Chemistry
and Materials Science**

- Topic interests and engages students because materials are familiar
- Unexpected material behaviors can lead to inquiry
 - Students hypothesize the answer to the question *How does this toy work?*
- Students design their own experiments to investigate additional questions they have

8/10/2002 T. Trupp

Mood Rings

- Physical Properties
 - Hollow glass shell filled with *liquid crystals*
 - Molecules change position (twist) according to the temperature
 - Increasing temperature: twists one direction
 - Decreasing temperature: twists in opposite direction
- Chemistry behind the mystery
 - Liquid crystals are *thermotropic*
 - Changes in structure are induced thermally

8/10/2002 T. Trupp

Silly Putty

- Physical Properties
 - stretches, bounces, shears under stress
 - *Viscoelastic* non-Newtonian fluid
- Chemistry behind the fun
 - Long chain polymer that is cross-linked
 - Belongs to a class known as *dilatants*
 - Has both viscous and elastic abilities
 - Chains can slip over each other, flowing like a liquid

8/10/2002 T. Trupp

Silly String

- Physical properties
 - Polymer foam stored under pressure
 - Materials used are an unknown industry secret
- Chemistry
 - Soluble in acetone?

Happy/Unhappy Balls

- Physical properties
 - Neoprene has elastic properties at room temperature, the other does not
- Chemistry
 - Neoprene rubber
 - Elasticity is due to flexibility of bonds joining the network
 - Norsorex rubber
 - Has weak bonds resulting in a low melting point
 - Absorbs most impacts

Memory Gels "Glueballs"

- Physical properties
 - Polymer that demonstrates the properties of both solids and liquids
 - Returns to original shape when stretched
 - similar to memory alloys
- Chemistry
 - Unknown at this time

Additional Explorations

- Mood rings
 - Can color be matched with temperature?
- Silly putty
 - Does viscosity change with temperature?
- Silly string
 - Does the polymer behave the same as styrofoam in acetone?
- Happy/Unhappy balls
 - Will a change in temperature affect behavior of material?
- Memory gels
 - How far can they stretch before breaking?
 - Will temperature affect "memory"?

Biopolymers

- Polymers produced naturally through biochemical reactions
- Biodegradable
- Feature oxygen or nitrogen atoms in their polymer backbones that are responsible for their biodegradability
 - Synthetic polymers have only carbon-carbon single bonds in their backbones (polyolefins) which resist biodegradation
- Produced by extracting carbohydrates, proteins from plant or animal material
 - Can also be made using microorganisms

Homemade Biopolymers

- Main components used
 - Starch (plant polysaccharide)
 - Gelatin (animal protein)
 - Agar (marine polysaccharide)
 - Plasticizers
 - Sorbitol
 - 1% Glycerol solution
 - > Increases the flexibility of the material
 - > Process is not completely understood
 - > Thought to form cross-links with starch

Discrepant Demonstration

- Used to engage students
- Styrofoam vs. Cornstarch packing peanuts
 - Leads to discussions about what materials each are made of
 - Inquiry about why each behaves differently
 - Can be used to introduce experimental design challenge

8/16/2002 T. Trapp

Design Challenge

- Requires students to act like scientists to meet specifications of product
- Specifications
 - Product must be flexible enough to wrap around a cardboard tube
 - Product must have the tensile strength to withstand 110 g
 - Product must be a minimum size of 8 X 10 cm²
- Constraints
 - Product must be within cost limitations

8/16/2002 T. Trapp

Process

- **Experimental research**
 - Create three different mixtures without plasticizer
 - Internet research
 - biopolymers (what they are and their importance), the materials used (what kind of chemicals, where they come from), plasticizers (what they are)
- **Development of final product**
 - Students decide (based on previous experimental research) what their product recipe will be
- **Experimental tests**
 - Must have evidence that product met specifications using digital pictures
- **Product proposal**
 - Students pitch their product to the "company" using Powerpoint

8/16/2002 T. Trapp

Experimental Tests on the Product

- Qualitative
 - Clarity
 - Flexibility at refrigeration temperatures
 - Permeability using vanilla extract
- Quantitative
 - Tensile strength
 - Glass temperature
 - Determines temperature range where flexibility is compromised
 - Dart test
 - Measures impact strength
 - Force of impact can be determined using a simple calculation

8/16/2002 T. Trapp

Further Research

- Can similar results be obtained by modifying the recipe?
- Substitute materials
 - Soy protein
 - Casein (milk protein)
 - Carrageenan (marine polysaccharide)
 - Aloe vera
 - Changing ratio of the materials used

8/16/2002 T. Trapp

Introductory Results

- Material was very flexible
- Texture differs from synthetic films
- Less tensile strength than synthetic

8/16/2002 T. Trapp

Place in Curriculum

- Toy Investigation
 - Relates to matter
 - Concepts can be reinforced in other units throughout the year
- Design Challenge
 - Organic chemistry
 - Environmental chemistry

Learning Outcomes

- Chemistry Standards
 - Explain the properties of materials in terms of the arrangement and properties of the atoms that compose them
 - Relate how the structure and arrangement of particles and their interactions determine the physical state of a substance at a given temperature and pressure
- MST Standards
 - **Standard 1:** use of scientific inquiry and engineering design to pose questions, seek answers, and develop solutions
 - **Standard 5:** Apply technological knowledge/skills to design, construct, use and evaluate products and systems to satisfy human and environmental needs.
 - **Standard 7:** Apply knowledge and thinking skills to address real-life problems and make informed decisions

Acknowledgements

- Dr. Geoff Coates for looking up the chemicals I needed and providing space for me to work on my biopolymer project
 - Also Yutan Getzler and all of the graduate students in Baker 472 who offered assistance and answered my questions
- Nev Singhota for helping me contact Dr. Coates and for ordering the chemicals I needed
- Ivan Johnson for loaning me his awesome iBook
- Dave Botsch for helping me work on various projects (other than my presentation!)
- Julianne and the rest of the staff for all of their help
- Anne, Mona, Randal, and Jerry for being great to work with and valuable resources

References

- Ellis, Arthur, B., et. al. Teaching General Chemistry: A Materials Science Companion. The American Chemical Society, 1993.
- Wilson, E. S. Green Plastics. Princeton University Press, 2002.

Web References

- Brain, Marshall. How Do Mood Rings Work? How Stuff Works. August 15, 2002. <<http://www.howstuffworks.com>>
- Yu, Jessen. How does Silly Putty work? Science Bug Articles. August 15, 2002. <<http://www.drawgroup.org/~jyu/writing/SciBug/SillyPutty.html>>
- Shape Memory Alloys. August 15, 2002. <<http://www.physics.pomona.edu/faculty/prof/TANENBAU/phys185/Smart%20Materials%20Site/Pages/Shape%20Memory%20Alloy.html>>
- Matejka, Libor. Polymer Networks. Fakulta Chemical. August 15, 2002 <<http://www.fch.vutbr.cz/mol/abomik/matej/>>
- Virtual Textbook: Chemical Properties of Liquid Crystals. Case Western Reserve University. August 15, 2002 <<http://abalone.cwu.edu/tutorial/enhanced/files/index.html>>
- YesterdayLand: Silly String. About.com. August 15, 2002. <<http://inventors.about.com/gldynamic/offsite.htm?site=http://www.yesterdayland.com/popopedia/shows/toys/ty1069.php>>